



Catalytic and selective conversion of (*Z*)-2-en-4-ynoic acids to either 2*H*-pyran-2-ones in the presence of ZnBr₂ or (*Z*)-5-alkylidenefuran-2(5*H*)-ones in the presence of Ag₂CO₃

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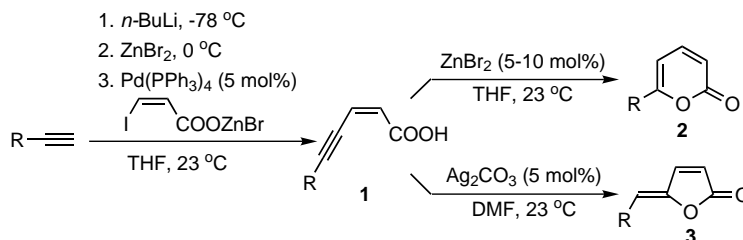
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Abstract—Treatment of (*Z*)-5-alkyl-2-en-4-ynoic acids (**1**), prepared by the Pd-catalyzed alkynylzinc-β-haloacrylic acid coupling, with 5–10 mol% of ZnBr₂ can produce 6-alkyl-2*H*-pyran-2-ones (**2**) along with minor amounts of (*Z*)-5-alkylidenefuran-2(5*H*)-ones (**3**) in >90% combined yields, with often very high (≈95/5) pyranone/furanone ratios. On the other hand, lactonization of **1** catalyzed by Ag₂CO₃ provides a selective synthesis of (*Z*)-5-alkylidenefuran-2(5*H*)-ones (**3**) in >90% yields along with minor amounts (≤5%) of **2**. © 2002 Published by Elsevier Science Ltd.

We report herein that treatment of 5-alkyl-2-en-4-ynoic acids (**1**) with 5–10 mol% of ZnBr₂ in THF can produce 6-alkyl-2*H*-pyran-2-ones (**2**) in high yields (Scheme 1). The pyranone/furanone ratio of approximately 95/5 have been observed with **1** containing *n*-hexyl, isobutyl, and cyclohexyl groups bonded to the C5 atom (entries 1–3 in Table 1). The use of DMF in place of THF in entry 3 lowered the **2**/**3** ratio to 67/33, although the combined yield was 91%. Cyclization of **1** to produce **2** in low yields (≤21%) along with the formation of (*Z*)-5-alkylidenefuran-2(5*H*)-ones (**3**) in 18–62% yields in the presence of a catalytic amount of AgI or Ag or 1 equiv. of HgO is known.¹ In the same publication, *o*-1-pentynylbenzoic acid was preferentially converted to 3-propylisocoumarin in good yields in the presence of catalytic amounts of AgClO₄, AgNO₃, AgOTf, or Ag₂CO₃. However, (*Z*)-2-en-4-ynoic acids (**1**) have not previously been catalytically converted to **2** in high yields as the major products.

During our study reported herein, conversion of **1** to **2** in low selectivity (**2**/**3** <4) by iodolactonization with 3 equiv. each of I₂ and NaHCO₃^{2a} followed by consecutive treatment with Zn and HCl was reported.^{2b} As the results shown in Scheme 2 indicate, the current procedure is superior to that reported previously in that it is a ZnBr₂-catalyzed one-step reaction of higher product yields and higher selectivity.

We further report that the preparation of (*Z*)-5-organyl-2-en-4-ynoic acids by the Pd-catalyzed alkynylzinc-β-haloacrylic acid coupling^{3–5} followed by lactonization catalyzed by Ag₂CO₃ provides a superior, albeit two-step, alternative to the recently developed one-pot alkyne-alkene coupling–lactonization tandem process^{6–8} run under Sonogashira coupling conditions⁹ for selective synthesis of (*Z*)-5-alkylidenefuran-2(5*H*)-ones (**3**) (Table 2).¹⁰



Scheme 1.

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Table 1. Selective conversion of (*Z*)-2-en-4-ynoic acids to 2*H*-pyran-2-ones catalyzed by ZnBr₂ in THF

entry	alkyne	yield of 1	time (h)	2 : 3	combined yield (%)	
					based on 1	based on alkyne
1	<i>n</i> -Hex—≡	95	24	94 : 6	95	90
2	<i>i</i> -Pr—≡	94	24	96 : 4	98	92
3	cyclohexyl—≡	93	48	96 : 4	92	86
4	<i>t</i> -Bu—≡	92	72	66 : 34	93	86
5	Ph—≡	94	72	50 : 50	94	88

Thus, in those cases where the R groups of the starting alkynes are *n*-Hex, *i*-Pr, cyclohexyl, *t*-Bu, and Ph, the overall yield of **3** are 82–88% over the two steps. These results favorably compare with the yield range of 46–74% for **3** observed in the one-pot tandem process. Furthermore, whereas the **3/2** ratio observed in the one-pot tandem process for those cases where the R groups are *n*-Hex, *i*-Pr, and cyclohexyl are only 2–4, those observed in the Ag-catalyzed lactonization are 19–22. These comparative results are summarized in Table 2. Some of the reported difficulties associated with the one-pot Pd-catalyzed cross-coupling-lactonization tandem process,^{6,7} such as alkyne dimerization and Michael-type addition reactions have not been observed in the two-step process.

Although *o*-alkynylbenzoic acids have been converted to dihydrobenzofuranone derivatives in high yields in the presence of Ag₂CO₃, AgI, or Ag,¹ AgI-catalyzed conversion of 5-phenyl-2-penten-4-ynoic acid into the corresponding **3**¹ and AgNO₃-catalyzed synthesis of TBS-protected lissoclinolide¹⁰ are the only previously reported examples of the Ag-catalyzed high-yield con-

version of **1** into **3**. In cases where R is alkyl, however, we find that AgNO₃ and a few of other Ag salts, such as AgOTf and AgBF₄, are unsatisfactory for selectively converting **1** into **3**. On the other hand, the use of Ag₂CO₃ or AgOAc, namely carbonate or carboxylate salts of Ag, leads to selective synthesis of **3** (Table 3).

On the basis of all of the findings presented above, it now appears reasonable to state that (i) the preparation of (*Z*)-2-en-4-ynoic acids (**1**) by the Pd-catalyzed cou-

Table 2. Comparison of Ag-catalyzed lactonization of (*Z*)-2-en-4-ynoic acids with Pd-catalyzed alkyne–alkene cross-coupling–lactonization tandem process

entry	alkyne	procedure ^a	2 : 3	yield of 2 + 3 (%)	
				based on 1	based on alkyne
1A	<i>n</i> -Hex—≡	A	4 : 96	96	90
1B	"	B	34 : 66	^b	77
2A	<i>i</i> -Pr—≡	A	5 : 95	99	93
2B	"	B	20 : 80	^b	75
3A	cyclohexyl—≡	A	3 : 97	95	88
3B	"	B	20 : 80	^b	65
4A	<i>t</i> -Bu—≡	A	2 : 98	95	86
4B	"	B	3 : 97	^b	76
5A	Ph—≡	A	2 : 98	94	88
5B	"	B	^c	^b	46 ^d

^aProcedure **A**: (1) Compound **1** was prepared under the cross-coupling conditions described in Ref. 11; (2) Ag₂CO₃ (5–10 mol%), DMF, 23°C, 1 h; Procedure **B**: CuI (5–10 mol%), Et₃N (4 equiv.), Pd(PPh₃)₄ (5 mol%), MeCN, 23°C, 12 h. ^bThe formation of compound **1** was never observed under conditions **B**. ^cRatio not determined. ^dRef. 6.

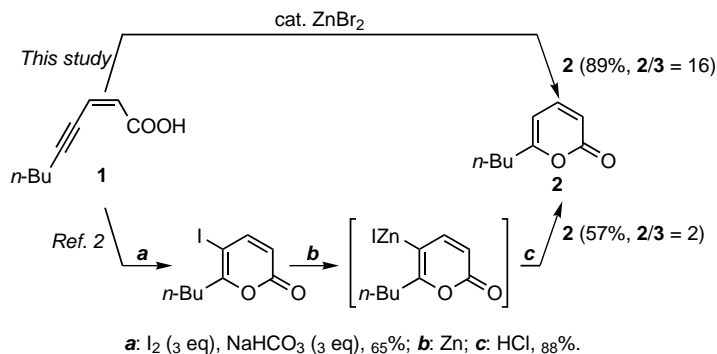

**Scheme 2.**

Table 3. The effects of Ag catalysts and solvents in the Ag-catalyzed lactonization of (*Z*)-2-en-4-ynoic acids

entry	alkyne	catalyst	solvent	2:3	combined yield (%) ^a
1	<i>n</i> -Hex \equiv	Ag ₂ CO ₃	DMF	4:96	96
2	"	AgOAc	DMF	9:91	97
3	"	AgNO ₃	DMF	64:36	95
4	"	AgNO ₃	MeOH	89:11	96
5	"	AgOTf	DMF	80:20	93
6	"	AgBF ₄	DMF	88:12	93
7		Ag ₂ CO ₃	DMF	3:97	95
8	"	AgOAc	MeCN	4:96	95
9	"	AgOAc	DMF	8:92	97
10	"	AgOAc	MeOH	9:91	92
11	"	AgNO ₃	MeOH	57:43	93
12	"	AgOTf	DMF	88:48	93

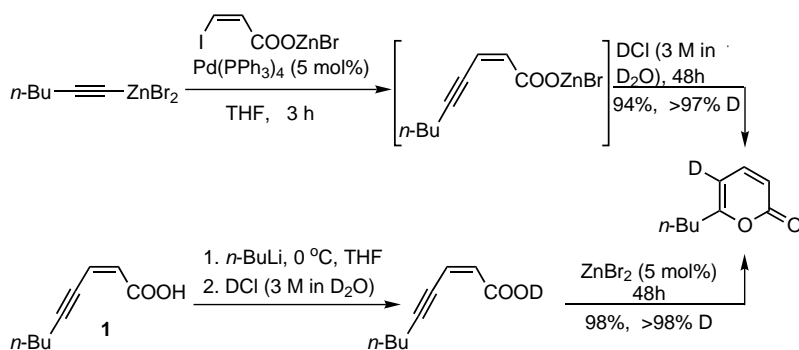
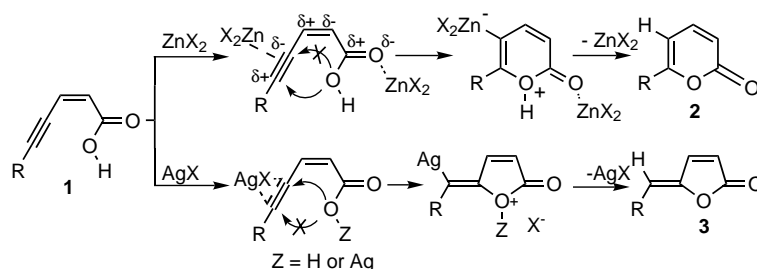
^aCombined yield of **2** and **3** based on **1**.

pling of alkynylzincs with (*Z*)- β -haloacrylic acids, (ii) the ZnBr₂-catalyzed cyclization of **1** to produce 2*H*-pyran-2-ones (**2**),¹¹ and (iii) the Ag₂CO₃-catalyzed cyclization of **1** to give (*Z*)-5-alkylidene-furan-2(5*H*)-ones (**3**),¹² all combine to provide an unprecedentedly

general and selective protocol for the synthesis of **2** and **3** from terminal alkynes and (*Z*)- β -haloacrylic acids (Scheme 1).^{13,14}

In our hands, Pd catalysts, such as Pd(PPh₃)₄, have not been effective in converting preformed and purified (*Z*)-2-en-4-ynoic acids into **2** and/or **3**. Only a very slow cyclization to give **2** and **3**, which can take place at comparable rates even in the absence of a Pd complex, has been observed. This point, however, is under further investigation. Interestingly, the zinc salts of (*Z*)-2-en-4-ynoic acids cannot be lactonized under the influence of ZnBr₂ either. However, addition of 3 M DCl in D₂O (1 mol equiv.) to the zinc salt of (*Z*)-2-nonen-4-ynoic acid induced a slow but smooth and quantitative lactonization with incorporation of D at the C4 position to the extent of >98%, indicating the intermediacy of the corresponding zinc derivative of 2*H*-pyran-2-one (Scheme 3).

At this point, factors affecting the pyranone/furanone ratio are not yet very clear. It is nevertheless tempting to speculate that polarization of the unsaturated bonds in (*Z*)-2-en-4-ynoic acids would favor the formation of 2*H*-pyran-2-ones via a Lewis acid-catalyzed polar process, whereas transition metal-catalyzed processes, in which bond polarization is less important, may stereoelectronically favor a 5-*exo*-mode cyclization (Scheme 4). Irrespective of the precise mechanisms of these cyclization reactions, however, the selective conversion of **1** into **2** and **3** catalyzed by ZnBr₂ and Ag₂CO₃, respectively, significantly complement a few related lactonization protocols developed over the past decade.

**Scheme 3.****Scheme 4.**

Acknowledgements

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- A representative procedure for the synthesis of **1** from the corresponding terminal alkyne is as follows: A solution of EtMgBr (1 M solution in THF, 2.1 mL) in 5 mL of THF was successively treated with ZnBr₂ (1 M solution in THF, 2.1 mL, freshly prepared from flame-dried ZnBr₂) and (Z)-3-iodo-2-propenoic acid (3.96 g, 2 mmol) in 2 mL of THF at 0°C under an Ar atmosphere. In a second flask a solution of a terminal alkyne (2 mmol) in 5 mL of THF was treated with *n*-BuLi (2.5 M solution in hexanes, 2.2 mL) at –78°C under an Ar atmosphere. After 10 min, the reaction mixture was treated with ZnBr₂ (1 M solution in THF, 2.2 mL), warmed to 0°C, transferred by cannula into the first flask, treated with Pd(PPh₃)₄ (115 mg, 0.1 mmol), and then warmed to 23°C. After 2–5 h the reaction mixture was diluted with Et₂O, washed with aqueous NH₄Cl and then with aqueous NaHCO₃, and treated with 1N NaOH. The aqueous layer was washed with hexanes, acidified with 1N HCl, extracted with Et₂O and dried over MgSO₄ to give the desired (Z)-2-en-4-ynoic acid (**1**). No further purification was necessary.
- A representative procedure for the metal-catalyzed lactonization of **1** to give **2** or **3** is as follows: To a solution of **1** (0.25 mmol) in 2–5 mL of solvent indicated in Tables 1–3 was added an appropriate metal salt (5–10 mol%) at 23°C. After the completion of the reaction indicated by TLC analysis, the reaction mixture was then filtered through a short path silica gel column and concentrated at reduced pressure. The products ratios and yields were determined by ¹H NMR spectroscopy and GC analyses using mesitylene as an internal standard. Chromatography on silica gel (95/5 hexane/EtOAc, v/v) allowed the separation of **2** and **3** and their isolation as pure substrates.
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